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- [8] Correct analytical values were obtained for the new compounds.

The Synthesis of Unsymmetrical Tetraarylporphyrins on Solid Phases^[**]

By Clifford C. Leznoff and Polina I. Svirskaya[*]

There has been tremendous interest recently directed toward the synthesis of model porphyrin systems related to the oxygen-binding proteins^[1]. In addition, the synthesis of model dimers and higher aggregates of porphyrins are of particular interest in studies of the photo-oxidation of water to oxygen during photosynthesis^[2]. The symmetrical tetraarylporphyrins had long been widely used as models in both of these systems due to their ease of preparation^[3], but some more recent sophisticated work required the synthesis of unsymmetrical tetraarylporphyrins, which were synthesized in yields ranging from 0.5—5%^[4]. Furthermore, the desired product could only be isolated from the many other compounds produced by extensive chromatographic separations.

Although preformed hemes and porphyrin models have been attached to insoluble supports^[5] and unusual synthetic polymers incorporating the heme function have been described^[6], the actual synthesis of a porphyrin on polymer supports from its simple precursors has not been described. Insoluble polymer supports provide a suitable means of "fishing out" a minor component from a complex reaction mixture^[7] and we have utilized these supports in this way to prepare monoblocked derivatives of a wide variety of symmetrical difunctional compounds^[7].

We now report that insoluble polymer supports can be used to prepare unsymmetrical tetraarylporphyrins and aid in their isolation. Thus a 2% crosslinked divinylbenzene-styrene copolymer (1) containing 1.7 mmol of benzoyl chloride functional groups per g polymer^[8], was treated with 3-hydroxybenzaldehyde (2) or 4-hydroxybenzaldehyde (3) to give the polymer-bound benzoylbenzaldehydes (4) and (5), respectively. Cleavage of (4) and (5) with 0.5 M sodium hydroxide/ dioxane (1:1) at room temperature for 20h liberated $0.73 \, \text{mmol}$ of (2) and $1.06 \, \text{mmol}$ of (3)/g of (4) and (5), respectively. Treatment of (4) and (5) with p-tolualdehyde and pyrrole in hot propionic acid for 1 h yielded upon filtration black polymers which were readily washed free of tetratolylporphyrin by extraction in a Soxhlet extractor with chloroform for 2 h to yield the polymer-bound unsymmetrical tetraarylporphyrins (6) and (7), respectively. Cleavage of (6) and (7) with potassium carbonate in methanol for 24 h yielded black powders essentially free of tetratolylporphyrin present in large amounts in a comparable solution synthesis[4a].

The black powders can be readily purified by a quick filtration through a silica column using chloroform as eluant followed by preparative thin-layer chromatography using benzene/ether/chloroform (7:2:1) as eluant to give 5-(3-hydroxyphenyl)-10,15,20-tritolylporphyrin (8) in 2% yield and 5-(4-hydroxyphenyl)-10,15,20-tritolylporphyrin (9) in 4.5% yield,

CHO

P

CHO

(2)

(1),
$$X = C1$$

HO

CHO

(3)

(4), $X = -0$

CHO

CHO

(5), $X = -0$

CHO

both as purple crystals. The NMR spectra of (8) and (9) were identical with the published spectra^[4a] and the mass spectra exhibited peaks of 672 (M^+) and 336 (M^{2^+}) typical of tetraarylporphyrins^[9].

CAS Registry numbers:

(2), 100-83-4; (3), 123-08-0; (8), 57412-06-3; (9), 57412-08-5; p-tolualdehyde, 104-87-0; pyrrole, 109-97-7

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Addition of Water to o-Phenylethynyl-Substituted Triphenylphosphanes

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Surprisingly, on recrystallization of the o-phenylethynylsubstituted triphenylphosphane (1a) 11a from commercial ethanol, the triphenylphosphane oxide (2a) containing a stereo-

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